

**catena-Poly[[triphenyltin(IV)]- $\mu$ -6-oxo-1,6-dihydropyridine-3-carboxylato]**

**Zhong-Jun Gao, Han-Dong Yin,\*  
 Gang Li and Da-Qi Wang**

College of Chemistry and Chemical Engineering,  
 Liaocheng University, Shandong 252059,  
 People's Republic of China

Correspondence e-mail: handongyin@163.com

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**Key indicators**

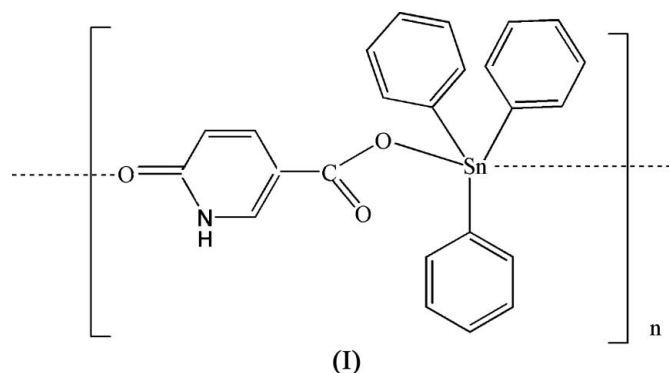
Single-crystal X-ray study  
 $T = 298$  K  
 Mean  $\sigma(\text{C}-\text{C}) = 0.006$  Å  
 $R$  factor = 0.027  
 $wR$  factor = 0.071  
 Data-to-parameter ratio = 14.3

For details of how these key indicators were  
 automatically derived from the article, see  
<http://journals.iucr.org/e>.

The title compound,  $[\text{Sn}(\text{C}_6\text{H}_5)_3(\text{C}_6\text{H}_4\text{NO}_3)]_n$ , possesses an infinite chain structure. The  $\text{SnO}_2\text{C}_3$  centre has distorted trigonal-bipyramidal geometry with the O atoms in the apical positions. A strong intermolecular  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bond results in the formation of double chains.

**Comment**

The title compound, (I) (Fig. 1), possesses an infinite one-dimensional chain structure arising from  $\text{Sn}-\text{O}$  bridges to the 6-hydroxy-3-pyridinecarboxylate ligand, one of which is substantially longer than the other (Table 1).

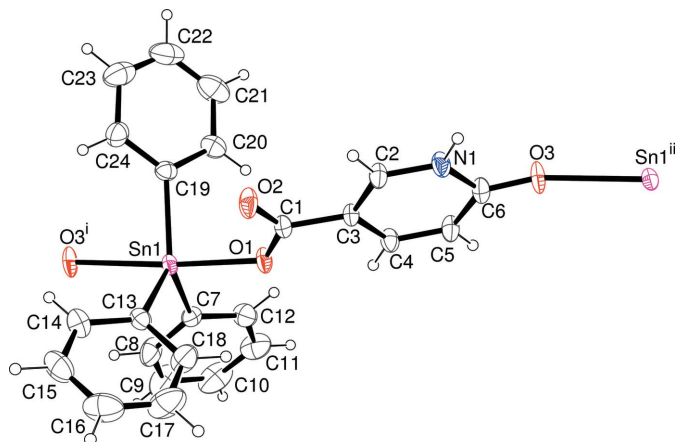


The Sn atom has distorted trigonal-bipyramidal geometry, with atoms O1 and O3<sup>i</sup> [symmetry code: (i)  $x, y - 1, z$ ] in axial positions [ $\text{O}1-\text{Sn}1-\text{O}3^i = 175.31(9)^\circ$ ] and the C atoms of the three phenyl groups in equatorial positions. The sum of the equatorial C-Sn-C angles is  $359.2^\circ$ , indicating approximate coplanarity for these atoms. The  $\text{SnO}_2\text{C}_3$  geometry in (I) is similar to those seen previously in related compounds (Xie *et al.*, 1991).

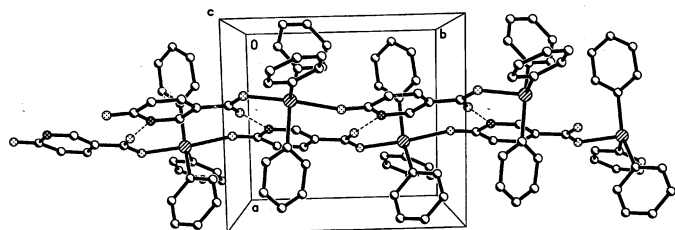
A strong intermolecular  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bond (Table 2) between the NH group of the pyridine ring and the non-coordinated O2 atom of a nearby carboxylate group results in the formation of a double chain parallel to  $b$  (Fig. 2).

**Experimental**

A mixture of triphenyltin oxide (1.4322 g, 2.0 mmol) and 6-hydroxynicotinic acid (0.5564 g, 4.0 mmol) in methanol (80 ml) was heated under reflux for 8 h. The resulting clear solution was evaporated under vacuum. The product was crystallized from a mixture of dichloromethane/ethanol (1:1) to yield blocks of (I). Yield 1.3862 g, 71%, m.p. 483 K. Analysis calculated for  $\text{C}_{24}\text{H}_{19}\text{NO}_3\text{Sn}$ : C 59.05, H 3.92, N 2.87%; found: C 59.02, H 3.96, N 2.91%.



**Figure 1**  
The molecular structure of (I), with 30% probability displacement ellipsoids (arbitrary spheres for H atoms). [Symmetry codes: (i)  $x, y - 1, z$ ; (ii)  $x, y + 1, z$ .]



**Figure 2**  
Part of a double chain in (I), with the  $N \cdots O$  hydrogen-bond contacts shown by dashed lines. H atoms have been omitted for clarity.

#### Crystal data

[ $\text{Sn}(\text{C}_6\text{H}_5)_3(\text{C}_6\text{H}_4\text{NO}_3)$ ]  
 $M_r = 488.09$   
 Monoclinic,  $P2_1/c$   
 $a = 9.5629$  (17) Å  
 $b = 10.6579$  (19) Å  
 $c = 21.353$  (4) Å  
 $\beta = 101.155$  (3)°  
 $V = 2135.2$  (7) Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.518$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 5143 reflections  
 $\theta = 2.2\text{--}27.8^\circ$   
 $\mu = 1.22$  mm<sup>-1</sup>  
 $T = 298$  (2) K  
 Block, colourless  
 $0.49 \times 0.46 \times 0.41$  mm

#### Data collection

Bruker SMART CCD diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 1998)  
 $T_{\min} = 0.586$ ,  $T_{\max} = 0.635$   
 10839 measured reflections

3755 independent reflections  
 3053 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.041$   
 $\theta_{\text{max}} = 25.0^\circ$   
 $h = -9 \rightarrow 11$   
 $k = -12 \rightarrow 12$   
 $l = -24 \rightarrow 25$

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.027$   
 $wR(F^2) = 0.071$   
 $S = 1.00$   
 3755 reflections  
 262 parameters  
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.034P)^2 + 0.6145P]$$

where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 0.52 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.47 \text{ e \AA}^{-3}$

**Table 1**

Selected geometric parameters (Å, °).

Sn1—C19	2.133 (3)	Sn1—O3 <sup>i</sup>	2.356 (2)
Sn1—C13	2.134 (3)	C1—O1	1.292 (4)
Sn1—C7	2.137 (3)	C1—O2	1.233 (4)
Sn1—O1	2.150 (2)		
C19—Sn1—C13	130.80 (12)	C7—Sn1—O1	90.01 (10)
C19—Sn1—C7	115.76 (12)	C19—Sn1—O3 <sup>i</sup>	83.52 (11)
C13—Sn1—C7	112.63 (12)	C13—Sn1—O3 <sup>i</sup>	87.84 (11)
C19—Sn1—O1	92.09 (10)	C7—Sn1—O3 <sup>i</sup>	90.42 (10)
C13—Sn1—O1	96.30 (11)	O1—Sn1—O3 <sup>i</sup>	175.31 (9)

Symmetry code: (i)  $x, y - 1, z$ .

**Table 2**

Hydrogen-bond geometry (Å, °).

$D\text{---}H \cdots A$	$D\text{---}H$	$H \cdots A$	$D \cdots A$	$D\text{---}H \cdots A$
N1—H1 <sup>ii</sup> ...O2 <sup>ii</sup>	0.86	1.95	2.785 (4)	165

Symmetry code: (ii)  $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$ .

H atoms were positioned geometrically [ $N\text{---}H = 0.86$  Å and  $C\text{---}H = 0.93$  Å] and refined as riding with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$ .

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1998); software used to prepare material for publication: SHELXTL.

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